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Transformations of 1-(2-Chloropyridyl-3)-4-ethoxycarbonyland 1-(2-Chloropyridyl-3)-4-ethoxycarbonylmethyl Thiosemicarbazides. Attempts to Prepare Pyrido[3,2-e]-1,2,4-thiadiazine

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2-Chloro-3-hydrazinopyridine (2) was converted with ethoxycarbonyl and ethoxycarbonylmethyl isothiocyanates into 1,4-disubstituted thiosemicarbazides 3 and 4, while with phenyl isothiocyanate directly 1*H*-pyrido[3,2-e]-1,3,4thiadiazine 7 was formed. Attempts to cyclize the thiosemicarbazides 3 and 4 into pyridothiadiazine derivatives 5 and 6 failed. In the reaction of 3 with hydrazine 2aminothiazolo[5,4-b]pyridine (9) was formed, while 4 gave only the corresponding hydrazide 10. The cyclization of the side chain occured in compound 4 by heating in aqueous hydrochloric acid to give 11, which was further transformed with N,N-dimethylformamide dimethyl acetal (*DMFDMA*) into 12, while with diethyl acetylene dicarboxylate the thiazolidone derivative 13 was produced.

(Keywords: 1,4-Disubstituted thiosemicarbazides; 1H-Pyrido[3,2-e]-1,3,4-thiadiazines; Thiazolinones)

Transformationen von 1-(2-Chlorpyridyl-3)-4-ethoxycarbonyl- und 1-(2-Chlorpyridyl-3)-4-ethoxycarbonylmethylthiosemicarbaziden. Versuche zur Synthese von Pyrido[3,2-e]-1,2,4-thiadiazinen

2-Chlor-3-hydrazinopyridin (2) wurde mit Ethoxycarbonyl- und Ethoxycarbonylmethylisothiocyanaten zu 1,4-disubstituierten Thiosemicarbaziden 3 und 4 umgesetzt, mit Phenylisothiocyanaten wurden direkt die 1*H*-Pyrido[3,2-e]-1,3,4thiadiazine 7 erhalten. Versuche zur Cyclisierung der Thiosemicarbazide 3 und 4 zu den Pyridothiadiazinderivaten 5 und 6 gelangen nicht. Bei der Reaktion von 3 mit Hydrazin entstand 2-Aminothiazolo[5,4-b]pyridin (9), 4 gab nur das entsprechende Hydrazid 10. Die Cyclisierung der Seitenkette von 4 gelang durch Erhitzen in wäßriger HCl unter Bildung von 11, das seinerseits mit N,N-Dimethylformamiddimethylacetal (*DMFDMA*) weiter zu 12 umgesetzt wurde, währenddessen mit Diethylacetylendicarboxylat die Thiazolidonderivate 13 entstanden.

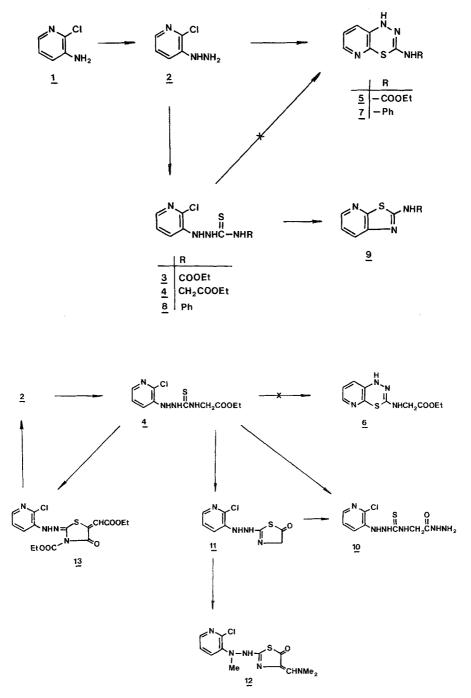
Introduction

The synthesis and reactions of condensed 1,3,4-thiadiazine systems have been of recent interest to a number of research groups [1-3]. The first synthesis of pyrido[3,2-e]-1,3,4-thiadiazine system has been acchieved by condensing N'-phenylbenzothiohydrazide and 2-chloro-3nitropyridine [1], while an attempted synthesis to prepare derivatives of the isomeric pyrido [2,3-e]-1,3,4-thiadiazine system was unsuccessful [2]. that 4-ethoxycarbonyl-1-(3-chloropyrazinyl-We have observed 2)thiosemicarbazide and 4-ethoxycarbonylmethyl-1-(3-chloropyrazinyl-2)thiosemicarbazide, prepared from 2-chloro-3-hydrazinopyrazine and ethoxycarbonylmethyl isothiocvanate and isoethoxycarbonyl thiocyanate, respectively, cyclized easily to give derivatives of the pyrazino[2,3-e]-1,3,4-thiadiazine system [3].

Results and Discussion

In the light of the above observation we have selected 2-chloro-3hydrazinopyridine (2) as the starting compound for the synthesis of pyrido[3,2-e]-1,3,4-thiadiazine derivatives. In the reaction of 2 with ethoxycarbonyl isothiocyanate or ethoxycarbonylmethyl isothiocyanate the corresponding 4-ethoxycarbonyl- (3) and 4-ethoxycarbonylmethyl-1-(2-chloropyridyl-3)thiosemicarbazide (4) were formed, which did not cyclize into pyrido[3,2-e]-1,3,4-thiadiazine derivatives 5 and 6, respectively. On the other hand, when 2 was heated in the presence of phenyl isothiocyanate in methanol a rapid conversion into 3-phenylamino-1Hpyrido[3,2-e]-1,3,4-thiadiazine (7) occured without isolation of 4-phenyl-1-(2-chloropyridyl-3)thiosemicarbazide (8) as an intermediate.

Some other transformations of thiosemicarbazides 3 and 4 were also performed. When 3 was heated with hydrazine hydrate in refluxing methanol for two hours, 2-aminothiazolo 5,4-b pyridine 9 was obtained in 61% yield, identical with the compound prepared form 2ethoxycarbonylthiazolo[5,4-b]pyridine by alkaline hydrolysis [4]. On the other hand, the thisemicarbazide 4 gave with hydrazine hydrate under essentially the same reaction conditions only the corre-4-carbazoylmethyl-1-(2-chloropyridyl-3)thiosemicarbazide sponding (10), while on heating in aqueous hydrochloric acid at reflux for one hour cyclization of the side chain took place to give 1-(2-chloropyridyl-3)-2-(5-oxothiazolinyl-2)hydrazine (11). Methylation under neutral conditions with N.N-dimethylformamide dimethyl acetal (DMFDMA) produced 1-[methyl-(2-chloropyridyl-3)]-2-(4-dimethylaminomethylidene-5-oxothiazolinyl-2)hydrazine (12), indicating that Nmethylation of one of the NH groups-as already observed in some other



24 Monatshefte für Chemie, Vol. 119/3

examples [5-11]—and transformation of an active methylene group into the dimethylaminomethylidene group—as observed previously in the pyridazine series [12]—had occured. The position of the N-methylation was established on the basis of a long-range coupling constant ³J: The coupling between ${}^{13}C_3$ of the pyridine ring ${}^{13}C$ —H and protons of the methyl group indicates that the methyl group must be attached at the nitrogen atom next to the pyridine ring.

By further treatment of 11 with hydrazine hydrate at room temperature the thiazolinone ring opened resulting in the formation of the hydrazide 10.

4 reacted with diethyl acetylenedicarboxylate to give 2-(2-chloropyridyl-3)-hydrazono-3-ethoxycarbonylmethyl-5-ethoxycarbonyl-methylidene-4-thiazolidone (13).

Experimental

Melting points were determined on a *Kofler* hot plate m.p. apparatus. ¹H NMR spectra were recorded on a JEOL C 60 HL spectrometer and ¹³C NMR spectra on a JEOL FX 90 Q spectrometer (*TMS* as internal standard, δ -values in ppm) and mass spectra on a Hitachi-Perkin-Elmer RMU-6L mass spectrometer. Elemental analyses (C, H, N) were obtained with a Perkin-Elmer Analyser 240 C.

2-Chloro-3-hydrazinopyridine (2)

To a solution of 128 mg 1 in 3 ml of aqueous hydrochloric acid (1:1) a solution of 69 mg of sodium nitrite in 1 ml of water was added dropwise at 0 °C. To the resulting solution 250 mg of stannous chloride was added in small portions and the mixture was stirred at 0 °C for 2 h, and than extracted with diethyl ether (5 times, 10 ml each time). The combined extracts were dried over anhydrous sodium sulphate and evaporated to give 2. The compound is unstable and decomposes by attempted purification. It was used without further purification for the preparation of 3, 4 and 7.

1-(2-Chloropyridyl-3)-4-ethoxycarbonyl thiosemicarbazide (3)

This compound was prepared from **2** according to the general procedure described in Ref. [18] in 82% yield, m.p. 224–226 °C (from methanol). MS (*m*/e): 274 (*M*⁺). NMR (*DMSO-d*₆): 1.29 (t, CH₃CH₂), 4.26 (q, CH₃CH₂), 7.43 (dd, H5'), 8.04 (dd, H4'), 8.41 (dd, H6'), 12.13 (br. s, NH), $J_{H4',H5'} = 8.5$ Hz, $J_{H4',H6'} = 1.5$ Hz, $J_{H5',H6'} = 5.4$ Hz, $J_{CH_3CH_2} = 6.9$ Hz.

 $C_{9}H_{11}ClN_{4}O_{2}S (274.72). Calcd. C 39.34 H 4.03 N 20.39. Found C 39.58 H 4.09 N 20.38.$

1-(2-Chloropyridyl-3)-4-ethoxycarbonylmethyl thiosemicarbazide (4)

This compound was prepared from 2 according to the general procedure described in Ref. [18] in 69% yield, m.p. 194–195 °C (from methanol). MS (m/e) 288 (M^+). NMR ($DMSO-d_6$): 1.19 (t, CH₃CH₂), 4.07 (q, CH₃CH₂), 4.19 (d, NHCH₂), 7.06 (dd, H4'), 7.33 (dd, H5'), 7.84 (dd, H6'), 8.13 (br. s, NH), 8.40 (t,

NHCH₂), 9.63 (br. s, NH), $J_{\text{H5',H6'}} = 4.6 \text{ Hz}$, $J_{\text{H4',H5'}} = 8.1 \text{ Hz}$, $J_{\text{H4',H6'}} = 1.6 \text{ Hz}$, $J_{\text{CH}_3\text{CH}_2} = 7.1 \text{ Hz}$, $J_{\text{NHCH}_2} = 5.4 \text{ Hz}$. $C_{10}\text{H}_{13}\text{Cln}_4\text{O}_2\text{S}$ (288.65). Calcd. C41.59 H4.53 N19.40.

Found C41.50 H4.46 N19.35.

3-Phenylamino-1H-pyrido[3,2-e]1,3,4-thiadiazine (7)

To a mixture of 143 mg 2 in 3 ml of methanol 160 mg of phenylisothiocyanate was added and the mixture was heated under reflux for 30 min. The precipitate was, after cooling, collected by suction to give 73 mg (30%) of the product, m.p. 173–175 °C (dec.) (from ethanol). MS (m/e): 242 (M^+), NMR ($DMSO-d_6$): 7.14 (m, H7,H8, 3-Ph), 7.76 (dd, H6), 9.70 (br. s, NH), 9.86 (br. s, NH), $J_{H6,H7} = 4.4$ Hz, $J_{H6,H8} = 1.6$ Hz.

 $\begin{array}{c} C_{12}H_{10}N_4S \ (242.29). \\ Found \ C \ 59.48 \ H \ 4.16 \ N \ 23.12. \\ Found \ C \ 59.27 \ H \ 4.32 \ N \ 23.26. \end{array}$

2-Aminothiazolo[5,4-b]pyridine (9)

A mixture of 200 mg 3, 1 ml of 80% hydrazine hydrate and 5 ml of methanol was heated under reflux for 2 h. The solvent was evaporated *in vacuo* and 2 ml of water was added to the oily residue. The precipitate was collected by suction to give 67 mg (61%) of 9, m.p. $252-254 \degree C$ (from ethanol). Ref. [4] m.p. $252-254 \degree C$. The IR spectrum of the compound was identical with that of an authentic sample [4].

4-Carbazoylmethyl-1-(2-chloropyridyl-3)-thiosemicarbazide (10)

a) To a suspension of 288 mg 4 in 5 ml of ethanol 1 ml of 80% hydrazine hydrate was added and the mixture was heated under reflux for 5 h. The solvent was evaporated *in vacuo* and the dry residue recrystallized from ethanol to give 143 mg (52%) of the product, m.p. 206–208 °C. NMR (*DMSO-d*₆): 4.01 (d, NHCH₂), 4.13 (br. s, NH₂), 7.01 (dd, H₄), 7.25 (dd, H5), 7.75 (dd, H6), 7.97 (br. s, NH), 8.13 (t, NHCH₂), 8.80 (br. s, NH), 9.49 (br. s, NH), $J_{H4',H5'} = 8.1$ Hz, $J_{H4',H6'} = 1.6$ Hz, $J_{H5',H6'} = 4.1$ Hz, $J_{NHCH_2} = 5.2$ Hz.

$$C_8 H_{11} CIN_6 OS \ (274.74). \ \ Calcd. \ C \ 34.97 \ H \ 4.03 \ N \ 30.59. \\ Found \ C \ 35.35 \ H \ 4.01 \ N \ 31.02.$$

This compound cyclizes by heating into 11.

b) A mixture of 242 mg 11, 5 ml of ethanol and 1 ml of 80% hydrazine hydrate was left at room temperature for 14 h. The volatile components were evaporated *in vacuo* and the dry residue was washed with ethanol to give 180 ml (66%) of 10. The IR spectrum of it was identical with that of the compound described under a).

1-(2-Chloropyridyl-3)-2-(5-oxadiazolinyl-2)-hydrazine (11)

A mixture of 100 mg 4 and 2 ml of aqueous hydrochloric acid (1:1) was heated under reflux for 1 h. The precipitate was, after cooling, collected by suction to give 22 mg (26%) of the product, m.p. 235 °C (dec.) (from ethanol). MS (*m*/e): 242 (*M*⁺). NMR (*DMSO-d*₆): 4.27 (s, CH₂), 7.03 (dd, H4), 7.21 (dd, H5'), 7.80 (dd, H6'), 8.41 (br. s, NH), 10.35 (br. s, NH), $J_{H5',H6'} = 4.6$ Hz, $J_{H4',H5'} = 8.1$ Hz, $J_{H4',H6'} = 1.5$ Hz.

> $C_8H_7CIN_4OS$ (242.69). Calcd. C 39.59 H 2.90 N 23.08. Found C 40.05 H 3.33 N 23.18.

337

B. Koren et al.:

1-[Methyl-(2-chloropyridyl-3)]-2-(4-dimethylaminomethylidene-5oxothiazolinyl-2)-hydrazine (12)

A mixture of 242 mg 11 and 2 ml of N,N-dimethylformamide dimethyl acetal (DMFDMA) was heated under reflux for 30 min. The volatile components were evaporated in vacuo to give 89 mg (29%) of the product, m.p. 242-244 °C (from methanol). MS (m/e): 311 (M^+). ¹H NMR ($DMSO-d_6$): 2.49 (s, Me), 3.14 (s, Me), 3.25 (s, Me), 6.69 (dd, H4), 7.05 (s, CH), 7.24 (dd, H5), 7.87 (dd, H6), 8.79 (br. s, NH), $J_{H4',H5'} = 8.1$ Hz, $J_{H4',H6'} = 1.5$ Hz, $J_{H5',H6'} = 4.6$ Hz. ¹³C NMR (*DMSO-d*₆): 11.6 (q, N-*Me*); 45.6 (m, N-*Me*₂), 112.7 (d, C4'), 120.3 (m, C5'), 123.8 (dd, C4'), 134.7 (m, C3'), 139.4 (m), 139.6 (m), 139.7 (m) (C6', C2', CH), 147.4 (q, C2'), 165.9 (d, C5').

> $C_{12}H_{14}ClN_5OS$ (311.80). Calcd. C46.22 H4.52 N 22.46. Found C46.02 H4.80 N22.17.

2-(2-Chloropyridyl-3)-hydrazono-3-ethoxycarbonylmethyl-5ethoxycarbonylmethylidene-4-thiazolidone (13)

A mixture of 288 mg 4, 170 mg of diethyl acetylene dicarboxylate and 3 ml of ethanol was heated under reflux for 45 min. The solvent was evaporated in vacuo to give 176 mg (42%) of the product, m.p. $155-156 \degree C$ (from methanol). MS (m/e): 412 (M^+). NMR ($DMSO-\hat{d}_6$): 1.27 (t, CH_2Me), 1.34 (t, CH_2Me), 3.39 (s, CH_2), 4.31 (q, CH₂Me), 7.04 (s, CH), 7.56 (m, H4', H5'), 8.08 (dd, H6'), 10.83 (br. s, NH), $J_{\text{H5',H6'}} = 4.1 \text{ Hz}, J_{\text{H4',H6'}} = 1.5 \text{ Hz}, J_{\text{CH}_2Me} = 7.6 \text{ Hz}.$

> C₁₆H₁₇ClN₄O₅S (412.86). Calcd. C 46.54 H 4.15 N 13.57. Found C46.46 H4.21 N13.43.

References

- [1] Elliott AJ, Gibson MS (1980) J Org Chem 45: 3677
- [2] Ogura H (1981) Khim Geterotski Soed 867
 [3] Koren B, Stanovnik B, Tišler M (1985) Heterocycles 23: 913
- [4] Petrič A, Stanovnik B, Tišler M, Verček B (1978) Vestn Slov Kem Drus 25: 31; C.A. (1978) 89: 109200d
- [5] Stanovnik B, Tišler M, Hribar A, Barlin GB, Brown DJ (1981) Aust J Chem 34: 1729
- [6] Stanovnik B, Štimac A, Tišler M, Verček B (1981) Vestn Slov Kem Drus 28: 427; C.A. (1982) 96: 217744n
- [7] Stanovnik B, Koren B, Šteblaj M, Tišler M, Žmitek J (1982) Vestn Slov Kem Drus 29: 129; C.A. (1983) 98: 53811v
- [8] Stanovnik B, Štimac A, Tišler M, Verček B (1982) J Heterocyclic Chem 19: 577
- [9] Stanovnik B, Mirtič T, Koren B, Tišler M, Belčič B (1982) Vestn Slov Kem Drus 29: 331; C.A. (1983) 98: 215542n
- [10] Stanovnik B, Podergajs S, Tišler M, Verček B (1983) Vestn Slov Kem Drus 30: 39; C.A. (1983) 99: 122348j
- [11] Stanovnik B, Prhavc M, Koren B, Tišler M (1983) Vestn Slov Kem Drus 30: 459; C.A. (1984) 100: 103249a
- [12] Krbavčič A, Povše L, Stanovnik B (1983) Heterocycles 20: 2347

338

- [13] Gartner A, Koren B, Stanovnik B, Tišler M (1984) Vestn Slov Kem Drus 31:
 1; C.A. (1984) 101: 230451f
- [14] Stanovnik B, Bajt O, Belčič B, Koren B, Prhavc M, Štimac A, Tišler M (1984) Heterocycles 22: 1545
- [15] Stanovnik B, Svete J (1986) Vestn Slov Kem Drus 33: 353; C.A. (1987) 107: 39701 g
- [16] Urleb U, Stanovnik B, Stibilj V, Tišler M (1986) Heterocycles 24: 1899
- [17] Merslavič M, Stanovnik B, Tišler M (1986) Monatsh Chem 117: 221
- [18] Koren B, Stanovnik B, Tišler M (1988) Monatsh Chem 119: 83